

## 論文内容要旨

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学位論文題目	Preparation and Characterization of CuO <sub>x</sub> -based Photocatalyst and NiFeP-based Electrocatalyst for Water Splitting (水分解用CuO <sub>x</sub> 光触媒とNiFeP電極触媒の製作と特性評価)		
<p>内容要旨</p> <p>Water splitting for hydrogen evolution is a potential technique to solve the crises of energy shortage and environment pollution. In recent years, this research area has achieved a great progress by the persistent efforts of the researchers.</p> <p>In this thesis, we reported on two kinds of methods for overall water splitting, include photoelectrochemical water splitting and electrocatalytic water splitting. Solar energy conversion through photoelectrochemical water splitting by semiconductors has becoming one of the most effective ways to solve energy and environmental issues. Electrocatalytic water splitting can be regarded as a promising one for clean energy and renewable. Synthesis of stable and efficient electrocatalysts is paramount importance for hydrogen evolution reaction, and oxygen evolution reaction.</p> <p>Firstly, the CuO<sub>x</sub> composite photocatalyst were deposited on FTO transparent conducting glasses by alternating current magnetron reactive sputtering under different Ar/O<sub>2</sub> ratios. The advantage of this deposited method is that it can deposit a CuO<sub>x</sub> photocatalyst uniformly and rapidly with large scale. From the photoelectrochemical properties of these CuO<sub>x</sub> photocatalysts, it can be found that the CuO<sub>x</sub> photocatalyst with Ar/O<sub>2</sub> 30:7 provide a photocurrent density of -3.2 mA/cm<sup>2</sup> under a bias potential -0.5 V (vs. Ag/AgCl), which was found to be twice higher than that of Ar/O<sub>2</sub> with 30:5. A detailed characterization on the structure, morphology and electrochemical properties of these CuO<sub>x</sub> photocatalysts were carried out, and it is found that the improved PEC performance of CuO<sub>x</sub> photocatalyst with Ar/O<sub>2</sub> 30:7 attributed to the less defects in it, indicating that this Ar/O<sub>2</sub> 30:7 is an optimized condition for excellent CuO<sub>x</sub> photocatalyst fabrication.</p> <p>Secondly, a CuO<sub>x</sub>/WO<sub>3</sub> photocatalyst based on p-n heterojunction proposed as a highly performance and stable photocathode. The CuO<sub>x</sub>/WO<sub>3</sub> photocatalyst was deposited by magnetron reactive sputtering layer by layer, followed with slow rate annealing in O<sub>2</sub> ambient. This is an excellent method for high-quality and uniform composite thin-film deposition with large areas at a high growth rate. The optimized CuO<sub>x</sub>/WO<sub>3</sub> photocatalyst after slow rate annealing at 500 °C in O<sub>2</sub> provides an obviously enhanced photoinduced current density of -3.8 mA cm<sup>-2</sup> at a bias potential of -0.5 V (vs. Ag/AgCl), which value is 1.5 times higher than that of bared CuO<sub>x</sub>. This highly enhanced photoelectrochemical performance is</p>			

attributed to p-n heterojunction, which accelerates the photogenerated electrons and holes transfer to n-WO<sub>3</sub> and p-CuO<sub>x</sub>, thereby accelerate the separation of photogenerated carries. In addition, WO<sub>3</sub> layer covered on the surface of CuO<sub>x</sub> photocatalyst can improve the stability of Cu<sub>2</sub>O in electrolytes.

Lastly, a new highly active and cost-effective nanocomposite electrocatalyst Ni-Fe-P-FeMnO<sub>3</sub> supported on nickel foam has been designed and prepared by electroless composite deposition and rapid dealloying. The nanocomposite electrocatalyst Ni-Fe-P-FeMnO<sub>3</sub>/NF presents a remarkable electrocatalytic activity towards HER, which demands a quite low overpotential of 16.62 mV at -10 mA cm<sup>-2</sup> current density in 1 M KOH, which behaves better performance than Pt/C noble metal catalysts. Moreover, Ni-Fe-P-FeMnO<sub>3</sub>/NF electrocatalyst exhibits remarkable OER activity as well, which requires only 297 mV and 219 mV of overpotential at 50 mA cm<sup>-2</sup> and 10 mA cm<sup>-2</sup>, respectively. Evidently, its overall water splitting activity used as both anode and cathode performs a low overpotential of 1.63 V to reach 10 mA cm<sup>-2</sup>. The high performances of the obtained electrocatalyst are mainly due to the unique structure of nano perovskite FeMnO<sub>3</sub> as well as the synergistic effect between nickel/iron and phosphorus.